

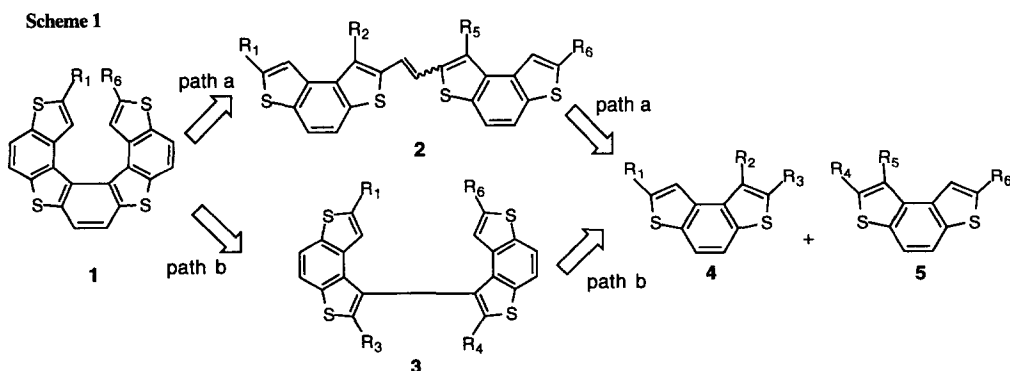
A Novel Route to Disubstituted [7]Thiaheterohelicene via Biaryl- and Carbonyl-Coupling Reactions

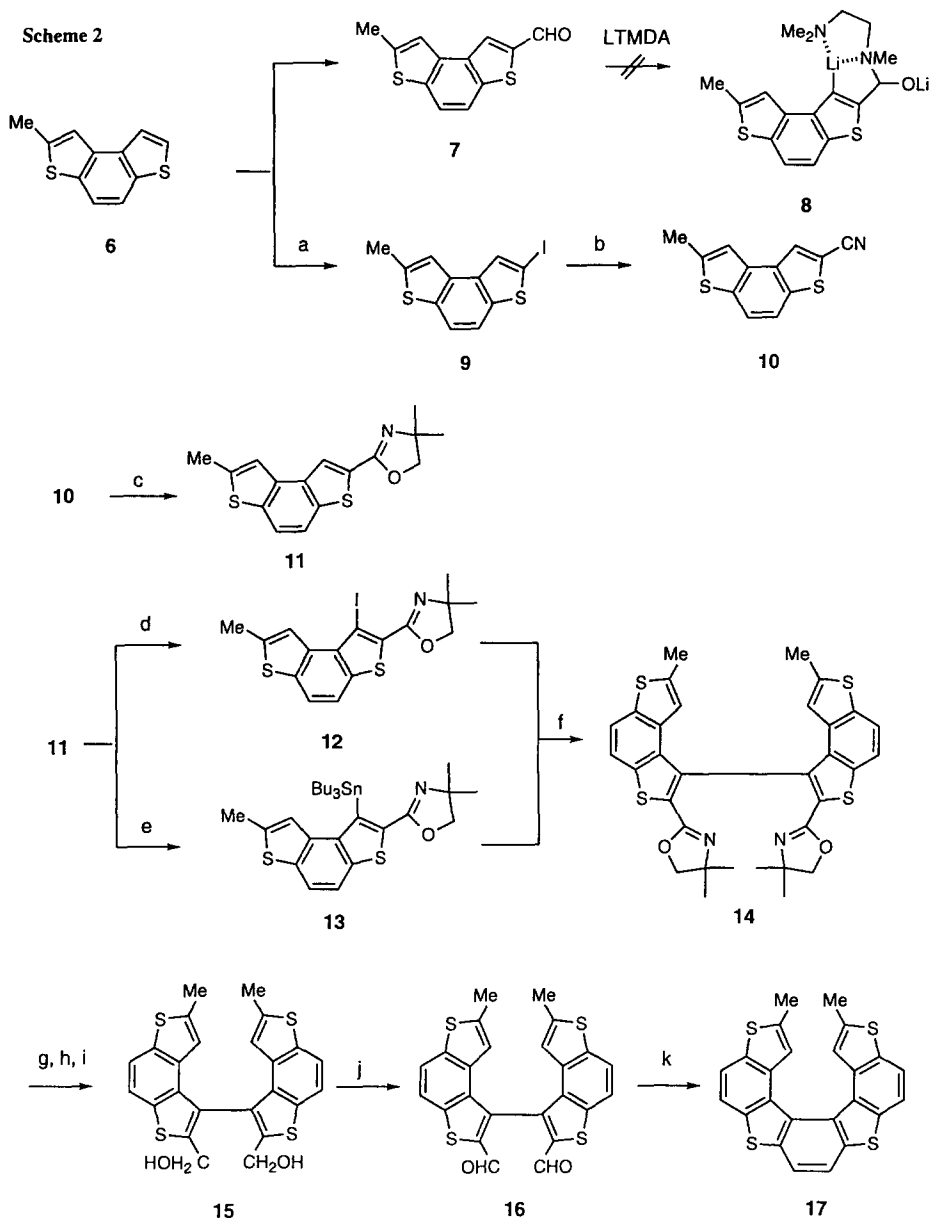
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Abstract: A new non-photochemical synthesis of disubstituted heterohelicene is described. The directed metalation at the 1-position of 2-(4,4-dimethyloxazolin-2-yl)-7-methylbenzo[1,2-b:4,3-b']dithiophene was accomplished by the use of BuLi in ether to give the organolithium species which, upon treatment with iodine and chlorotributyltin, gave 2-(4,4-dimethyloxazolin-2-yl)-1-iodo-7-methylbenzo[1,2-b:4,3-b']dithiophene and 2-(4,4-dimethyloxazolin-2-yl)-7-methyl-1-tributylstannylbenzo[1,2-b:4,3-b']dithiophene in high yields. Palladium-catalyzed cross-coupling between them, followed by the intramolecular McMurry coupling reaction, gave 2,13-dimethyl[7]thiaheterohelicene in good yield. Copyright © 1996 Elsevier Science Ltd

Functionalized chiral helicenes and their heterocyclic analogs are of considerable interest in both materials research¹ and molecular recognition² due to their very high specific optical rotations and unique non-planar π -electron structures based on helicity.³ In the synthesis of these overcrowding aromatic molecules, oxidative photocyclization is a crucial step, but the process has several drawbacks and limitations: the intense irradiation by UV light must be conducted in very dilute solutions (0.3-0.4 mM)⁴ and is not always compatible with the presence of sensitive functional groups.⁵ Although several non-photocyclization methods for helicenes have been reported, including the Friedel-Crafts acylation,⁶ the intramolecular oxidative cyclization,⁷ the cyclization of ammonium or phosphonium salts,⁸ the acid or base-catalyzed condensation reactions,^{9, 10, 11} and the Diels-Alder reactions of benzoquinone,¹² there is still a need for complementary methods which proceed under mild conditions and extend the range of application.





Reagents and conditions: (a) (i) BuLi, THF; (ii) iodine; (b) copper(I) cyanide, DMF, 74% from 6; (c) 2-amino-2-methyl-1-propanol, cat. zinc chloride, chlorobenzene, 95%; (d) (i) BuLi, ether; (ii) iodine, 95%; (e) (i) BuLi, ether; (ii) chlorotributyltin, 91%; (f) cat. PdCl₂(PPh₃)₂, THF, 62%; (g) TFA, sodium sulfate, THF, H₂O; (h) acetic anhydride, pyridine; (i) LiAlH₄, THF, 71%; (j) PCC, CH₂Cl₂, 94%; (k) TiCl₃·DME_{1.5}, Zn·Cu, DME, 78%.

This paper reports a non-photochemical route to disubstituted heterohelicene, which utilizes the palladium-catalyzed cross-coupling between two aryl units, **4** and **5**, and then the intramolecular McMurry coupling between the R₃ and R₄ substituents of the biaryl skeleton **3** (Scheme 1; path b). In the photochemical pathway (path a), the two aryl moieties are first connected at R₃ and R₄ by a vinylene bridge,⁵ and the resulting 1,2-diarylethylene **2** is subjected to cyclization between the R₂ and R₅ groups.

Our initial experiments aimed at the lithiation of the 3-position of 2-formyl-7-methylbenzo[1,2-b:4,3-b']dithiophene¹³ (**7**) using lithiated *N,N,N'*-trimethylethylenediamine (LTMDA)¹⁴ were unsuccessful and the starting material was recovered (Scheme 2). However, we observed that the introduction of a strong ortho-directing oxazoline moiety¹⁵ at the 7-position of 2-methylbenzo[1,2-b:4,3-b']dithiophene (**6**) led to a facile generation of the organolithium species. The oxazoline **11** was prepared from the corresponding cyanide **10**, which was readily available in 74% overall yield from **6**. Treatment of **11** with BuLi in ether at -78 °C, followed by trapping the lithium anion with iodine and chlorotributyltin, gave **12** and **13** in 95% and 91% yield, respectively.

The cross-coupling reaction¹⁶ between **12** and **13** smoothly proceeded in the presence of 5 mol% of PdCl₂(PPh₃)₂ to afford the biaryl compound **14** in 62% yield. The oxazoline moiety was then transformed to the corresponding dialdehyde **16** via oxidation of **15** by PPC.¹⁵ Finally, the intramolecular McMurry coupling reaction¹⁷ using TiCl₃/Zn-Cu gave 2,13-dimethyl[7]thiaheterohelicene **17** in 32% yield. The use of TiCl₃·DME_{1,5}/Zn-Cu¹⁸ increased the yield to 78%. The protons (δ 6.43) on C(1) and C(14) of **17** resonate at higher field than the proton (δ 7.32) on C(1) of **6**. This implies that the former protons are in the region of a shielding zone exerted by the helical aromatic structure.¹⁹

In summary, we have developed a practical method for disubstituted thiaheterohelicene using as the key steps the Stille coupling and the McMurry coupling reactions. The methyl groups of **17** can be easily converted to other functionalities such as the bromomethyl moiety¹³ and are thus very appropriate for further synthetic studies. If two different heterocycles having chiral oxazoline moieties could be subjected to the palladium-catalyzed cross-coupling, then a new practical advance in methodology for the asymmetric synthesis of helicenes should result. Work along these lines is in progress and will be reported in due course.

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References and Notes

- (a) Ashitaka, H.; Yokoh, Y.; Shimizu, R.; Yokozawa, T.; Morita, K.; Suehiro, T.; Matsumoto, Y. *Nonlinear Optics*, **1993**, *4*, 281-297; (b) Morita, K.; Suehiro, T.; Yokoh, Y.; Ashitaka, H. *J. Photopolymer Sci. Tec.* **1993**, *6*, 229-238.
- (a) Nakazaki, M.; Yamamoto, K.; Ikeda, T.; Kitsuki, T.; Okamoto, Y. *J. Chem. Soc., Chem. Commun.* **1983**, 787-788; (b) Yamamoto, K.; Ikeda, T.; Kitsuki, T.; Okamoto, Y.; Chikamatsu, H.; Nakazaki, M. *J. Chem. Soc., Perkin 1*, **1990**, 271-276; (c) Tanaka, K.; Osuga, H.; Shogase, Y.; Suzuki, H.; Nakanishi, W.; Nakamura, K.; Kawai, Y. *J. Chem. Soc., Chem. Commun.* **1995**, 1873-1874.

3. (a) Meurer, K. P.; Vögtle, F. *Top. Currn. Chem.* **1985**, *127*, 1-76; (b) Laarhoven, W. H.; Prinsen, W. J. C. *Top. Currn. Chem.* **1984**, *125*, 63-130; (c) Martin, R. H. *Angew. Chem., Int. Ed. Eng.* **1974**, *13*, 649-660; (d) Wynberg, H. *Acc. Chem. Res.* **1971**, *4*, 65-73.
4. Gilbert, A. M.; Katz, T. J.; Geiger, W. E.; Robben, M. P.; Rheingold, A. L. *J. Am. Chem. Soc.* **1993**, *115*, 3199-3211.
5. Mallory, F. B.; Mallory, C. W. *Organic Reactions*; Wiley & Sons: New York, 1984; Vol. 30, pp. 1-456.
6. Newman, M. S.; Lednicer, D. *J. Am. Chem. Soc.* **1956**, *78*, 4765-4770.
7. (a) Pereira, D. E.; Neelima; Leonard, N. J. *Tetrahedron* **1990**, *46*, 5895-5908; (b) Larsen J.; Bechgaard, K. *J. Org. Chem.* **1996**, *61*, 1151-1152; (c) Rau, H.; Schuster, O. *Angew. Chem., Int. Ed. Eng.* **1976**, *15*, 114-115.
8. (a) Bestmann, V. J.; Both, W. *Angew. Chem.* **1972**, *84*, 293; (b) Stará, I. G.; Star'ý, I.; Tich'ý, M.; Závada, J.; Hanus, V. *J. Am. Chem. Soc.* **1994**, *116*, 5084-5088.
9. Dore, A.; Fabbri, D.; Gladiali, S.; Valle, G. *Tetrahedron: Asymmetry* **1995**, *6*, 779-788.
10. Fritsch, R.; Hartmann, E.; Andert, D.; Mannschreck, A. *Chem. Ber.* **1992**, *125*, 849-855.
11. (a) Högberg, H.-E. *Acta Chem. Scand.* **1973**, *27*, 2591-2596; (b) Zander, M.; Franke, W. H. *Chem. Ber.* **1969**, *102*, 2728-2738; (c) Bogaert-Verhoogen, D.; Martin, R. H. *Tetrahedron Lett.* **1967**, 3045; (d) Teuber, H.-J. Vogel, L. *Chem. Ber.* **1970**, *103*, 3319-3342; (e) Bringmann, G.; Schöner, B.; Schupp, O.; Peters, K.; Peters, E.-M.; von Schnering, H. G. *Liebigs Ann. Chem.* **1994**, 91-97.
12. (a) Nuckolls, C.; Katz, T. J.; Castellanos, L. *J. Am. Chem. Soc.* **1996**, *118*, 3767-3768; (b) Willmore, N. D.; Hoic, D. A.; Katz, T. J. *J. Org. Chem.* **1994**, *59*, 1889-1891; (c) Liu, L.; Katz, T. J. *Tetrahedron Lett.* **1990**, *31*, 3983-3986, and references cited therein; (d) Willmore, N. D.; Liu, L.; Katz, T. J. *Angew. Chem., Int. Ed. Engl.* **1992**, *31*, 1093-1095.
13. Lehman, P. G.; Wynberg, H. *Aust. J. Chem.* **1974**, *27*, 315-322.
14. Comins, D. L.; Killpack, M. O. *J. Org. Chem.* **1987**, *52*, 104-109.
15. (a) Reuman, M.; Meyers, A. I. *Tetrahedron* **1985**, *41*, 837-860; (b) Vecchia, L. D.; Vlattas, I. *J. Org. Chem.* **1977**, *42*, 2649-2650.
16. (a) Stille, J. K. *Angew. Chem., Int. Ed. Engl.* **1986**, *25*, 508-524; (b) Crisp, G. T. *Synth. Commun.* **1989**, *19*, 307-316; (c) Gronowitz, G. *J. Heterocycl. Chem.* **1994**, *31*, 641-662.
17. (a) McMurry J. E. *Chem. Rev.* **1989**, *89*, 1513-1524; (b) Hu, Z.; Cava, M. P. *Tetrahedron Lett.* **1994**, *35*, 3493-3496; (c) Yoshida, S.; Fujii, M.; Aso, Y.; Otsubo, T.; Ogura, F. *J. Org. Chem.* **1994**, *59*, 3077-3081.
18. McMurry, J. E.; Lectka, T.; Rico, J. G. *J. Org. Chem.* **1989**, *54*, 3748-3749.
19. (a) Tanaka, K.; Kitahara, Y.; Suzuki, H.; Osuga, H.; Kawai Y. *Tetrahedron Lett.* **1996**, *37*, 5925-5928; (b) Nakagawa, H.; Obata, A.; Yamada, K.; Kawazura, H. *J. Chem. Soc., Perkin. Trans. 2*, **1985**, 1899-1903; (c) Yamada, K.; Ishii, R.; Nakagawa, H.; Kawazura, H. *Tetrahedron: Asymmetry* **1996**, *7*, 737-746.

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